



**SYNTHESIS AND BIOLOGICAL ACTIVITY OF PYRROLE AND PYRROLIDINE
COMPOUNDS FROM 4-BROMO-2-HYDROXYBENZOIC ACID HYDRAZIDE**



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Abstract

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Condensation of 4-bromo-2-hydroxy benzoic acid hydrazide (1) with aromatic aldehydes to gives the corresponding 4-bromo-2-hydroxy benzoic acid arylidene hydrazides (2a-h). Cyclocondensation of compounds 4-bromo-2-hydroxy benzoic acid arylidene hydrazides (2a-h) with maleic anhydride and succinic anhydride yields respectively 1-(4-bromo-2-hydroxybenzamido)-5-oxo-2-aryl-2,5-dihydro-1H-pyrrole-3-carboxylic acid (3a-h) and 1-(4-bromo-2-hydroxybenzamido)-5-oxo-2-arylpyrrolidine-3-carboxylic acid (4a-h). The structures of these compounds were established on the basis of analytical and spectral data. All the newly synthesized compounds were evaluated for their antibacterial and antifungal activities.

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Introduction

4-bromo-2-hydroxy benzoic acid hydrazide is heterocyclised product and the Hydrazides and their heterocyclised products display diverse biological activities including antibacterial, antifungal, analgesic, anti-inflammatory properties. These heterocyclic systems find wide use in medicine, agriculture and industry. One of the hydrazides, 4-bromo-2-hydroxy benzoic acid hydrazide and their condensed products play a vital role in medicinal chemistry [1-16]. Pyrrole and pyrrolidine compounds give good biological and pharmacological properties [17]. Hence, it was thought of interest to merge both of pyrrole and 4-bromo-2-hydroxy benzoic acid hydrazide, pyrrolidine and 4-bromo-2-hydroxy benzoic acid hydrazide moieties which may enhance the drug activity of compounds to some extent, or they might possess some of the above mentioned biological activities. From this point of view, the objective of the present work is to prepare new derivatives of 4-bromo-2-hydroxy benzoic acid hydrazide containing pyrrole and pyrrolidine moiety. Hence the present communication comprises the synthesis of 1-(4-bromo-2-

hydroxybenzamido)-5-oxo-2-aryl-2,5-dihydro-1H-pyrrole-3-carboxylic acid (3a-h) and 1-(4-bromo-2-hydroxybenzamido)-5-oxo-2-arylpyrrolidine-3-carboxylic acid (4a-h). The synthetic approach is shown in scheme-1.

Material and Method

Melting points were determined in open capillary tubes and were uncorrected. The IR spectra were recorded in KBr pellets on a Nicolet 400D spectrometer and ^1H NMR and ^{13}C NMR spectra were recorded in DMSO with TMS as internal standard on a Bruker spectrometer at 400 MHz and 100 MHz, respectively. LC-MS of selected samples taken on LC-MSD-Trap-SL_01046.

Preparation of 4-bromo-2-hydroxy benzoic acid arylidene hydrazide (2a-h)

General procedure: – An equimolecular mixture of 4-bromo-2-hydroxy benzoic acid hydrazide (1), (0.01mole) and the aromatic aldehydes (a-h) in ethanol (25mL) was refluxed on a water bath for 2h. The solid separated was collected by filtration, dried and recrystallized from ethanol. The yields, melting points and other characterization data of these compounds are given in Table -1.

Preparation of 1-(4-bromo-2-hydroxybenzamido)-5-oxo-2-aryl-2,5-dihydro-1H-pyrrole-3-carboxylic acid (3a-h)

General procedure: A mixture 4-bromo-2-hydroxy benzoic acid arylidene hydrazide (2a-h) (0.1 mole) and Maleic anhydride (0.1 mole) were heated at reflux in chloroform (50ml) for about 3.5 hours with TLC monitoring. After the mixture was allowed to stand for 1 days, the solid was filtered. The product thus formed was recrystallized from ethanol to give pure 1-[4-pyridinylcarbonylamino]-2-oxo-5-aryl-2,5-dihydro-1H-pyrrole-3-carboxylic acid in (3a-h), which were obtained in 58-61% yield.

The yields, melting points and other characterization data of these compounds are given in Table -2.

Preparation of 1-(4-bromo-2-hydroxybenzamido)-5-oxo-2-arylpyrrolidine-3-carboxylic acid (4a-h)

A mixture 4-bromo-2-hydroxy benzoic acid arylidene hydrazide (2a-h) (0.1 mole) and Succinic anhydride (0.1 mole) were heated at reflux in chloroform (50ml) for about 4 hours with TLC monitoring. After the mixture was allowed to stand for 1 day, the solid was filtered. The product thus

formed was recrystallized from ethanol to give pure 1-[4-pyridinylcarbonyl amine]-2-oxo-5-aryl-2,5-dihydro-1H-pyrrolidinone-3-carboxylic acid (3a-h) in good yield.

The yields, melting points and other characterization data of these compounds are given in Table -3.

Biological Screening

Antibacterial activities

The antibacterial activities of all the compounds were studied against gram-positive bacteria (*Staphylococcus aureus* and *Bacillus subtilis*) and gram-negative bacteria (*E.coli*, and *klebsiella promioe*) at a concentration of 50µg/mL by agar cup plate method. A methanol system was used as control in this method. Similar conditions using tetracycline as a control was used standard for comparison. The area of inhibition of zone measured in cm. Compounds 3f, 3h, 4g, and 4h were found more toxic for microbes. Other compounds found to be less or moderate active than tetracycline Tables 4 and 5.

Antifungal Activities

The fungicidal activity of all the compounds was studied at 1000 ppm concentration in

vitro. Plant pathogenic organisms used were *Nigrospora* Sp, *Aspergillus Niger*, *Botrydepladia thiobromine*, and *Rhizopus nigricum*, *Fusarium oxyporium*. The antifungal activity of all the compounds (3a-h) & (4a-h) were measured on each of these plant pathogenic strains on a potato dextrose agar (PDA) medium. Such a PDA medium contained potato 200g, dextrose 20g, agar 20g and water 1c. Five days old cultures were employed. The compounds to be tested were suspended (1000ppm) in a PDA medium and autoclaved at 120° C for 15 min. at 15atm. pressure. These media were poured into sterile Petri plates and the organisms were inoculated after cooling the Petri plates. The percentage inhibition for fungi was calculated after five days using the formula given below:

$$\text{Percentage of inhibition} = 100(X-Y) / X$$

Where, X = Area of colony in control plate

Y = Area of colony in test plate

The fungicidal activity displayed by various compounds (3a-h) and (4a-h) is shown in Tables-6 and 7.

Results and Discussion

It was observed that 4-bromo-2-hydroxy benzoic acid hydrazide (1), on condensation with aromatic aldehydes, yields 4-bromo-2-hydroxy benzoic acid arylidene hydrazides (2a-h). The structures of (2a-h) were confirmed by elemental analysis and IR spectra showing an absorption band at 1620-1640 (C=N), 3030-3080 cm^{-1} (C-H, of Ar.), 3240-3260 cm^{-1} (-OH), 2815, 1250 cm^{-1} (-OCH₃), 2950, 1370 cm^{-1} (-CH₃). ¹H NMR : 6.85-7.84 (8H, m) (Ar - H), 11.70-11.84 (1H, s) (-OH), 11.85-11.98 (1H, s) (-CONH), 8.36-8.80 (1H, s) (-N=CH), 2e; 2.39 (3H, s) (-CH₃), 2b, 2g; 3.89 (3H, s) (-OCH₃), 2h; 4.12 (4H, q) (CH₂), 1.34 (6H, t) (CH₃), 2f; 6.12 (2H, s) (-OCH₂O- cyclic). ¹³C NMR: 111.8-160.8 (Aromatic), 163.0-164 (-CONH), 146.3-147 (-CH); (2b,2g): 55.7-56.8 (-OCH₃); (2e): 21 (CH₃); (2f): 102.5 (OCH₂O cyclic); (2h): 65.4 (OCH₂), 15.2 (CH₃). The C, H, and N analysis data of all compounds are presented in Table -1.

The structures assigned to 1-(4-bromo-2-hydroxybenzamido)-5-oxo-2-aryl-2,5-dihydro-1H-pyrrole-3-carboxylic acid (3a-h) were supported by the elemental analysis and IR spectra showing an absorption bands at 1667-1690 cm^{-1} (C=O of -COOH), 1715-1720 (C=O of pyrrole-2-one), 3030-3080 cm^{-1}

¹ (C-H, of Ar.), 3240-3260 cm⁻¹ (-OH). ¹H NMR: 6.62-7.82 (8H, m) (Ar-H), 5.50-5.60 (1H, s) (-C₅H of the ring), 7.10-7.15 (1H, s) (-C₃H), 12.90-12.98 (1H, s) (-COOH), 11.70-11.85 (1H, s) (-OH), 3e; 2.36 (3H, s) (-CH₃), 3b,3g; 3.93,3.90 (3H, s) (-OCH₃), 3h; 4.13 (4H q) (CH₂), 1.32 (6H, t) (CH₃), 3f; 6.11 (2H, s) (-OCH₂O- cyclic). ¹³C NMR: 110-161 (Aromatic), 55.0-62.5 (-CH), 169.5-171.5 (-COOH), 164.5-165 (-CO of the ring), (3b,3g): 55.4-56.5 (-OCH₃); (3e): 21.5 (CH₃); (3f): 102.8 (OCH₂O cyclic); (3h): 65.3 (OCH₂), 14.6 (CH₃). The C, H, and N analysis data of all compounds are presented in Table-2.

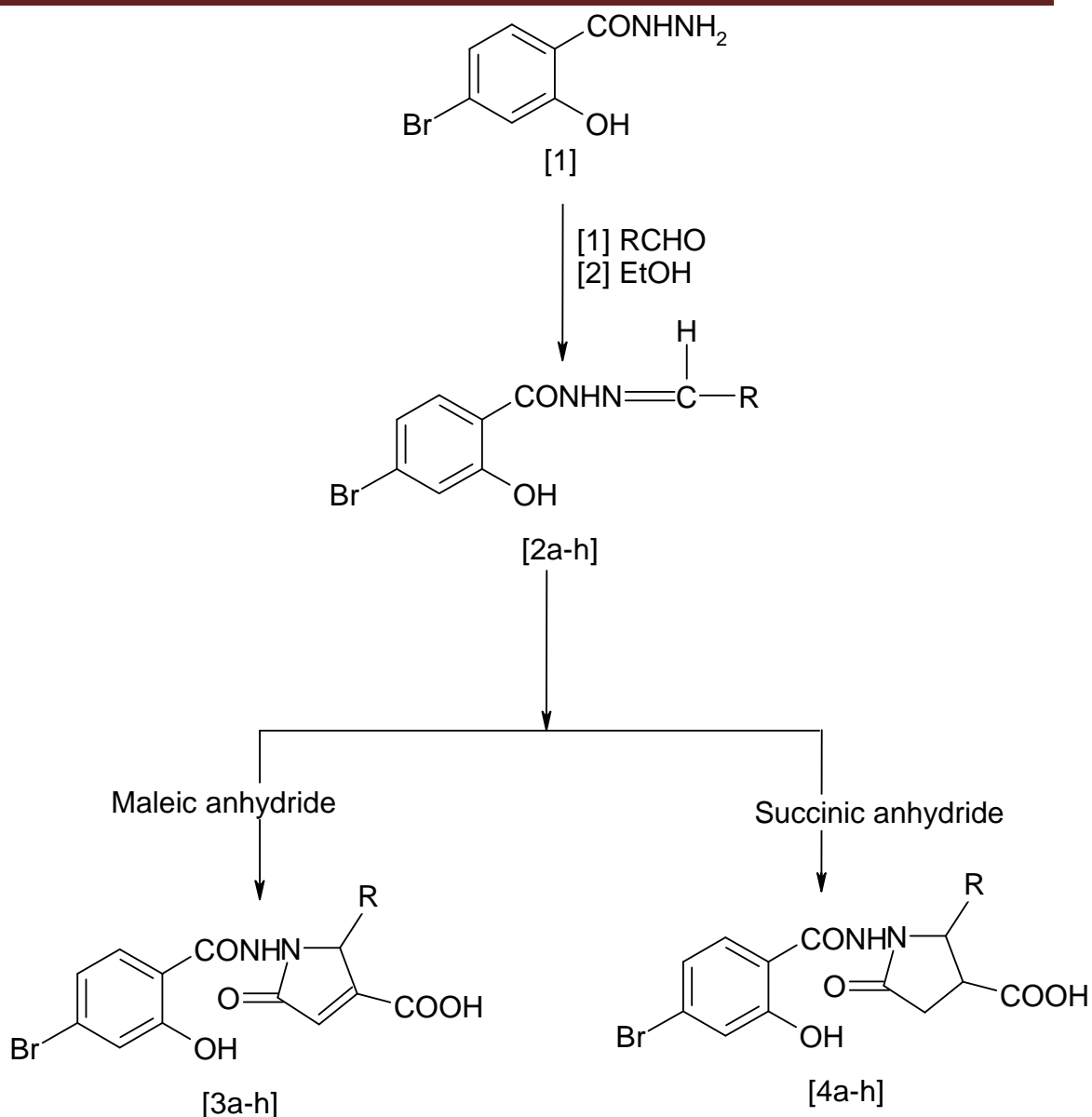
The structures assigned to 1-(4-bromo-2-hydroxybenzamido)-5-oxo-2-arylpyrrolidine-3-carboxylic acid (4a-h) were supported by the elemental analysis and IR spectra showing an absorption bands at 1667-1690cm⁻¹ (C=O of -COOH), 1715-1720 (C=O of pyrrole-2-one), 3030-3080 cm⁻¹ (C-H, of Ar.), 3240-3260 cm⁻¹ (-OH). ¹H NMR: 6.65-7.85 (8H, m) (Ar-H), 5.52-5.60 (1H, s) (-C₅H), 3.33-3.36 (1H, s) (-C₄H), , 2.50-2.54, 2.75-2.80 (2H, s) (-C₃H), 11.70-11.85 (1H, s) (-OH), 12.90-12.95 (1H, s) (-

COOH), 4e; 2.38 (3H, s) (-CH₃), 4b, 4g; 3.86 (3H, s) (-OCH₃), 4h; 4.06, (4H, q) (-CH₂), 1.35 (6H, t) (-CH₃), 4f; 6.08 (2H, s) (-OCH₂O cyclic). ¹³C NMR: 110-161 (Aromatic), 50.0-57.0 (-CH of the ring), 37.0-37.5 (-CH₂), 172.0-172.5 (-CO), 178.0-179 (-COOH), (4b,4g): 55.5-56.5 (-OCH₃); (4e): 21.6 (CH₃); (4f): 101.5 (OCH₂O); (4h): 65.3 (OCH₂), 15.0 (CH₃). The C, H, and N analysis data of all compounds are presented in Table-3.

The examination of elemental analytical data reveals that the elemental contents are consistence with the predicted structure shown in Scheme-1. The IR data also direct for assignment of the predicted structure. The final structure of all compounds is confirmed by LC-MS. LC-MS of 3c and 4f compounds are 443 and 471 respectively.

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SCHEME – 1

Where, R =(a) C₆H₅ (b) 4-OCH₃-C₆H₄ (c) 4-OH-C₆H₄ (d) 2-OH-C₆H₄

(e) 4-CH₃-C₆H₄ (f) 3,4-CH₂O₂-C₆H₄ (g) 4-OH-3-OCH₃-C₆H₃ (h) 3,4-C₂H₅-C₆H₄

Table:-1 Analytical Data and Elemental Analysis of Compounds (2a-h)

Compd.	Molecular formula (Mol.wt.)	Yield	M.P.* °C	Elemental Analysis					
				%C		%H		%N	
				Found	Calcd.	Found	Calcd.	Found	Calcd.
2a	C ₁₄ H ₁₁ BrN ₂ O ₂ (319)	75	248	52.55	52.66	3.39	3.44	8.70	8.77
2b	C ₁₅ H ₁₃ BrN ₂ O ₃ (349)	72	251	51.52	51.57	3.69	3.72	7.95	8.02
2c	C ₁₄ H ₁₁ BrN ₂ O ₃ (335)	76	252	50.03	50.14	3.15	3.28	8.29	8.35
2d	C ₁₄ H ₁₁ BrN ₂ O ₃ (335)	77	253	50.05	50.14	3.16	3.28	8.27	8.35
2e	C ₁₅ H ₁₃ BrN ₂ O ₂ (333)	81	249	53.69	53.73	3.82	3.90	8.33	8.40
2f	C ₁₅ H ₁₁ BrN ₂ O ₄ (363)	72	254	49.49	49.59	2.93	3.03	7.60	7.71
2g	C ₁₅ H ₁₃ BrN ₂ O ₃ (365)	71	257	49.21	49.31	3.51	3.56	7.59	7.67
2h	C ₁₈ H ₁₉ BrN ₂ O ₄ (407)	75	266	52.98	53.07	4.58	4.67	6.79	6.88

* Uncorrected

Table:-2 Analytical Data and Elemental Analysis of Compounds (3a-h)

Compd.	Molecular formula (Mol.wt.)	Yield	M.P.* °C	Elemental Analysis					
				%C		%H		%N	
				Found	Calcd.	Found	Calcd.	Found	Calcd.
3a	C ₁₈ H ₁₃ BrN ₂ O ₅ (417)	60	235	51.72	51.80	3.09	3.12	6.67	6.71
3b	C ₁₉ H ₁₅ BrN ₂ O ₆ (447)	58	231	50.94	51.01	3.29	3.36	6.21	6.26
3c	C ₁₈ H ₁₃ BrN ₂ O ₆ (433)	62	186	49.82	49.88	2.89	3.00	6.41	6.47
3d	C ₁₈ H ₁₃ BrN ₂ O ₆ (433)	61	188	49.81	49.88	2.89	3.00	6.40	6.47
3e	C ₁₉ H ₁₅ BrN ₂ O ₅ (431)	59	175	52.80	52.90	3.41	3.48	6.43	6.50
3f	C ₁₉ H ₁₃ BrN ₂ O ₇ (461)	63	184	49.39	49.46	2.76	2.82	5.98	6.07
3g	C ₁₉ H ₁₅ BrN ₂ O ₇ (463)	60	181	49.18	49.24	3.19	3.24	5.97	6.05
3h	C ₂₂ H ₂₁ BrN ₂ O ₇ (505)	62	228	52.18	52.28	4.09	4.16	5.49	5.54

* Uncorrected

Table:-3 Analytical Data and Elemental Analysis of Compounds (4a-h)

Compd.	Molecular formula (Mol.wt.)	Yield	M.P. °C	Elemental Analysis					
				%C		%H		%N	
				Found	Calcd.	Found	Calcd.	Found	Calcd.
4a	C ₁₈ H ₁₅ BrN ₂ O ₅ (419)	64	239	51.46	51.55	3.51	3.58	6.62	6.68
4b	C ₁₉ H ₁₇ BrN ₂ O ₆ (449)	60	241	50.69	50.78	3.71	3.79	6.22	6.24
4c	C ₁₈ H ₁₅ BrN ₂ O ₆ (435)	58	240	49.59	49.66	3.39	3.45	6.38	6.44
4d	C ₁₈ H ₁₅ BrN ₂ O ₆ (435)	63	237	49.60	49.66	3.37	3.45	6.37	6.44
4e	C ₁₉ H ₁₇ BrN ₂ O ₅ (433)	59	231	52.56	52.66	3.85	3.93	6.41	6.47
4f	C ₁₉ H ₁₅ BrN ₂ O ₇ (463)	60	239	49.18	49.24	3.20	3.24	5.98	6.05
4g	C ₁₉ H ₁₇ BrN ₂ O ₇ (465)	61	241	48.97	49.03	3.56	3.66	5.88	6.02
4h	C ₂₂ H ₂₃ BrN ₂ O ₇ (507)	59	239	52.01	52.07	4.45	4.54	5.43	5.52

* Uncorrected

Table:-4 Antibacterial Activity of Compounds (3a-h)

Compounds	Gram +Ve		Gram -Ve	
	Staphylococcus aureus	Bacillus subtilis	E.coli	Klebsiella promioe
3a	12	10	14	13
3b	11	14	12	18
3c	10	12	11	11
3d	13	14	14	13
3e	16	10	15	15
3f	15	17	13	19
3g	13	13	19	17
3h	18	19	16	15
Tetracycline	22	21	18	21

Table:-5 Antifungal Activity of Compounds (3a-h)

Zone of Inhibition at 1000 ppm (%)					
Compounds	Nigrospora	Aspergillus	Botrydepladia	Rhizopus	Fusarium
	Sp.	Niger	Thiobromine	Nigricum	oxyporium
3a	58	63	64	64	65
3b	59	61	65	65	68
3c	61	68	68	68	65
3d	62	62	60	71	68
3e	60	60	63	70	71
3f	62	61	67	63	69
3g	60	69	69	65	68
3h	63	71	70	72	69

Table:-6 Antibacterial Activity of Compounds (4a-h)

Compounds	Gram +Ve		Gram -Ve	
	Staphylococcus aureus	Bacillus subtilis	E.coli	Klebsiella promioe
4a	13	13	12	11
4b	15	15	15	17
4c	16	18	13	13
4d	13	14	18	15
4e	11	15	14	19
4f	16	16	14	18
4g	18	15	18	21
4h	16	16	15	18
Tetracycline	20	21	20	22

Table:-7 Antifungal Activity of Compounds (4a-h)

Zone of Inhibition at 1000 ppm (%)					
Compounds	Nigrospora	Aspergillus	Botrydepladia	Rhizopus	Fusarium
	Sp.	Niger	Thiobromine	Nigricum	oxyporium
4a	68	69	60	59	61
4b	65	65	63	61	63
4c	65	68	62	65	66
4d	62	61	62	62	60
4e	63	63	65	69	62
4f	60	59	67	66	67
4g	66	61	69	63	68
4h	62	63	71	72	70

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