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SYNTHESIS AND BIOLOGICAL ACTIVITY OF 1, 3, 5 - OXADIAZINE AND 1, 3, 4 - OXADIAZOLE COMPOUNDS FROM 4-CHLORO-2-HYDROXYBENZOIC ACID HYDRAZIDE



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Abstract

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4-chloro-2-hydroxy benzoic acid hydrazide (1) undergoes facile condensation with aromatic aldehydes to afford the corresponding 4-chloro-2-hydroxy benzoic acid arylidene hydrazides (2a-h) in good yields. Cyclocondensation of compounds (2a-h) with benzoyl isothiocyanate and acetic anhydride yields respectively of 4-chloro-2-Hydroxy – N – (2-Substituted Phenyl)-6-Phenyl-4-Thioxo-2H-1,3,5-Oxadiazin -3(4H)-yl Benzamide (3a-h) and 4-chloro-1-(5-(2-Hydroxyphenyl)-2-(Substituted Phenyl)-1,3,4-Oxadiazole-3) (2H-yl)ethane (4a-h). The structures of these compounds were established on the basis of analytical and spectral data. All the newly synthesized compounds were evaluated for their antibacterial and antifungal activities.

INTRODUCTION

Hydrazide and their heterocyclised products display diverse biological activities including antibacterial, antifungal, analgesic, anti-inflammatory properties [1-13]. These heterocyclic systems find wide use in medicine, agriculture and industry. One of the hydrazides, 4-chloro-2-hydroxy benzoic acid hydrazide and their condensed products play a vital role in medicinal chemistry [14-16]. 1,3,5-oxadiazine and 1,3,4-oxadiazole compounds give good biological and pharmacological properties [17-18]. Hence, it was thought of interest to merge both of 1,3,5-oxadiazine and 4-chloro-2-hydroxy benzoic acid hydrazide, 1,3,4-oxadiazole and 4-chloro-2-hydroxy benzoic acid hydrazide moieties which may enhance the drug activity of compounds to some extent, or they might possess some of the above mentioned biological activities. From this point of view, the objective of the present work is to prepare new derivatives of 4-chloro-2-hydroxy benzoic acid hydrazide containing 1,3,5-oxadiazine and 1,3,4-oxadiazole moiety. Hence the present communication comprises the synthesis of 4-chloro-2-Hydroxy - N - (2-Substituted Phenyl)-6-Phenyl-4-Thioxo-2H-1,3,5-

Oxadiazin -3(4H)-yl) Benzamide (3a-h) and 4-chloro-1-(5-(2-Hydroxyphenyl)-2-(Substituted Phenyl)-1,3,4-Oxadiazole-3(2H)-yl)ethane (4a-h). The synthetic approach is shown in scheme-1.

MATERIAL AND METHOD

Melting points were determined in open capillary tubes and were uncorrected. The IR spectra were recorded in KBr pellets on a Nicolet 400D spectrometer and ^1H NMR and ^{13}C NMR spectra were recorded in DMSO with TMS as internal standard on a Bruker spectrometer at 400 MHz and 100 MHz, respectively. LC-MS of selected samples taken on LC-MSD-Trap-SL_01046.

Preparation of 4-chloro-2-hydroxy benzoic acid arylidene hydrazide (2a-h)

General procedure: – An equimolecular mixture of 4-chloro-2-hydroxy benzoic acid hydrazide (1), (0.01mole) and the aromatic aldehydes (a-h) in ethanol (15mL) was refluxed on a water bath for 1-2 h. The solid separated was collected by filtration, dried and recrystallized from ethanol. The yields, melting points and other characterization data of these compounds are given in Table -1.

Preparation of 4-chloro-2-Hydroxy – N – (2-Substituted Phenyl)-6-Phenyl-4-Thioxo-2H-1,3,5-Oxadiazin -3(4H)-yl) Benzamide (3a-h)

General procedure: A mixture 4-chloro-2-hydroxy benzoic acid arylidene hydrazide (2a-h) (0.01 mole) benzoyl isothiocyanate and triethylamine (3 drops) in 1,4-dioxane (30mL) was heated at reflux for about 3h. The reported solid formed upon dilution with water (20mL) was filtered, dried and which were obtained in 57-68% yield.

The yields, melting points and other characterization data of these compounds are given in Table -2.

Preparation of 4-chloro-1-(5-(2-Hydroxyphenyl)-2-(Substituted Phenyl)-1,3,4-Oxadiazole-3) (2H)-yl)ethane (4a-h)

A mixture 4-chloro-2-hydroxy benzoic acid arylidene hydrazide (2a-h) (0.003 mole) and acetic anhydride 10mL were heated at reflux for about 4h. after the reaction mixture attained room temp. Excess acetic anhydride was decomposed by water poured in to ice cold water and the solid separated was recrystallized from ethanol to yield. Which were obtained in 55-65% yield.

The yields, melting points and other characterization data of these compounds are given in Table -3.

Biological Screening

Antibacterial activities

The antibacterial activities of all the compounds were studied against gram-positive bacteria (Staphylococcus aureus and Bacillus subtilis) and gram-negative bacteria (E.coli, and klebsiella promioe) at a concentration of 50µg/mL by agar cup plate method. A methanol system was used as control in this method. Similar conditions using tetracycline as a control was used standard for comparison. The area of inhibition of zone measured in cm. Compounds 3f, 3h, 4g, and 4h were found more toxic for microbes. Other compounds found to be less or moderate active than tetracycline Tables -4 and 5.

Antifungal Activities

The fungicidal activity of all the compounds was studied at 1000 ppm concentration in vitro. Plant pathogenic organisms used were Nigrospora Sp, Aspergillus niger, Botrydepladia thiobromine, and Rhizopus nigricum, Fusarium oxyporium. The

antifungal activity of all the compounds (3a-h) & (4a-h) were measured on each of these plant pathogenic strains on a potato dextrose agar (PDA) medium. Such a PDA medium contained potato 200g, dextrose 20g, agar 20g and water 1c. Five days old cultures were employed. The compounds to be tested were suspended (1000ppm) in a PDA medium and autoclaved at 120° C for 15 min. at 15atm. pressure. These media were poured into sterile Petri plates and the organisms were inoculated after cooling the Petri plates. The percentage inhibition for fungi was calculated after five days using the formula given below:

$$\text{Percentage of inhibition} = 100(X-Y) / X$$

Where, X = Area of colony in control plate

Y = Area of colony in test plate

The fungicidal activity displayed by various compounds (3a-h) and (4a-h) is shown in Tables-6 and 7.

RESULTS AND DISCUSSION

It was observed that 4-chloro-2-hydroxy benzoic acid hydrazide (1), on condensation with aromatic aldehydes, yields 4-chloro-2-hydroxy benzoic acid arylidene hydrazides (2a-h). The structures of (2a-h) were

confirmed by elemental analysis and IR spectra showing an absorption band at 1620-1640 (C=N), 3030-3080 cm⁻¹ (C-H, of Ar.), 3240-3260 cm⁻¹ (-OH), 2815, 1250 cm⁻¹ (-OCH₃), 2950, 1370 cm⁻¹ (-CH₃). ¹H NMR : 6.85-7.84 (8H, m) (Ar - H), 11.70-11.84 (1H, s) (-OH), 11.85-11.98 (1H, s) (-CONH), 8.36-8.80 (1H, s) (-N=CH), 2e; 2.39 (3H, s) (-CH₃), 2b, 2g; 3.89 (3H, s) (-OCH₃), 2h; 4.12 (4H, q) (CH₂), 1.34 (6H, t) (CH₃), 2f; 6.12 (2H, s) (-OCH₂O- cyclic). ¹³C NMR: 111.8-160.8 (Aromatic), 163.0-164 (-CONH), 146.3-147 (-CH); (2b,2g): 55.7-56.8 (-OCH₃); (2e): 21 (CH₃); (2f): 102.5 (OCH₂O cyclic); (2h): 65.4 (OCH₂), 15.2 (CH₃). The C, H, and N analysis data of all compounds are presented in Table -1.

The structures assigned to 1-(4-chloro-2-hydroxybenzamido)-5-oxo-2-aryl-2,5-dihydro-1H-pyrrole-3-carboxylic acid (3a-h) were supported by the elemental analysis and IR spectra showing an absorption bands at 1667-1690cm⁻¹ (C=O of -COOH), 1715-1720 (C=O of pyrrole-2-one), 3030-3080 cm⁻¹ (C-H, of Ar.), 3240-3260 cm⁻¹ (-OH). ¹H NMR: 6.62-7.82 (8H, m) (Ar-H), 5.50-5.60 (1H, s) (-C₅H of the ring), 7.10-7.15 (1H, s) (-C₃H), 5.60-5.68 (1H-s) (CH-oxadiazine), 12.90-12.98 (1H, s) (-COOH), 11.70-11.85

(1H, s) (-OH), 3e; 2.36 (3H, s) (-CH₃), 3b,3g; 3.93,3.90 (3H, s) (-OCH₃), 3h; 4.13 (4H q) (CH₂), 1.32 (6H, t) (CH₃), 3f; 6.11 (2H, s) (-OCH₂O- cyclic). ¹³C NMR: 110-161 (Aromatic), 55.0-62.5 (-CH), 169.5-171.5 (-COOH), 164.5-165 (-CO of the ring), (3b,3g): 55.4-56.5 (-OCH₃); (3e): 21.5 (CH₃); (3f): 102.8 (OCH₂O cyclic); (3h): 65.3 (OCH₂), 14.6 (CH₃). The C, H, and N analysis data of all compounds are presented in Table-2.

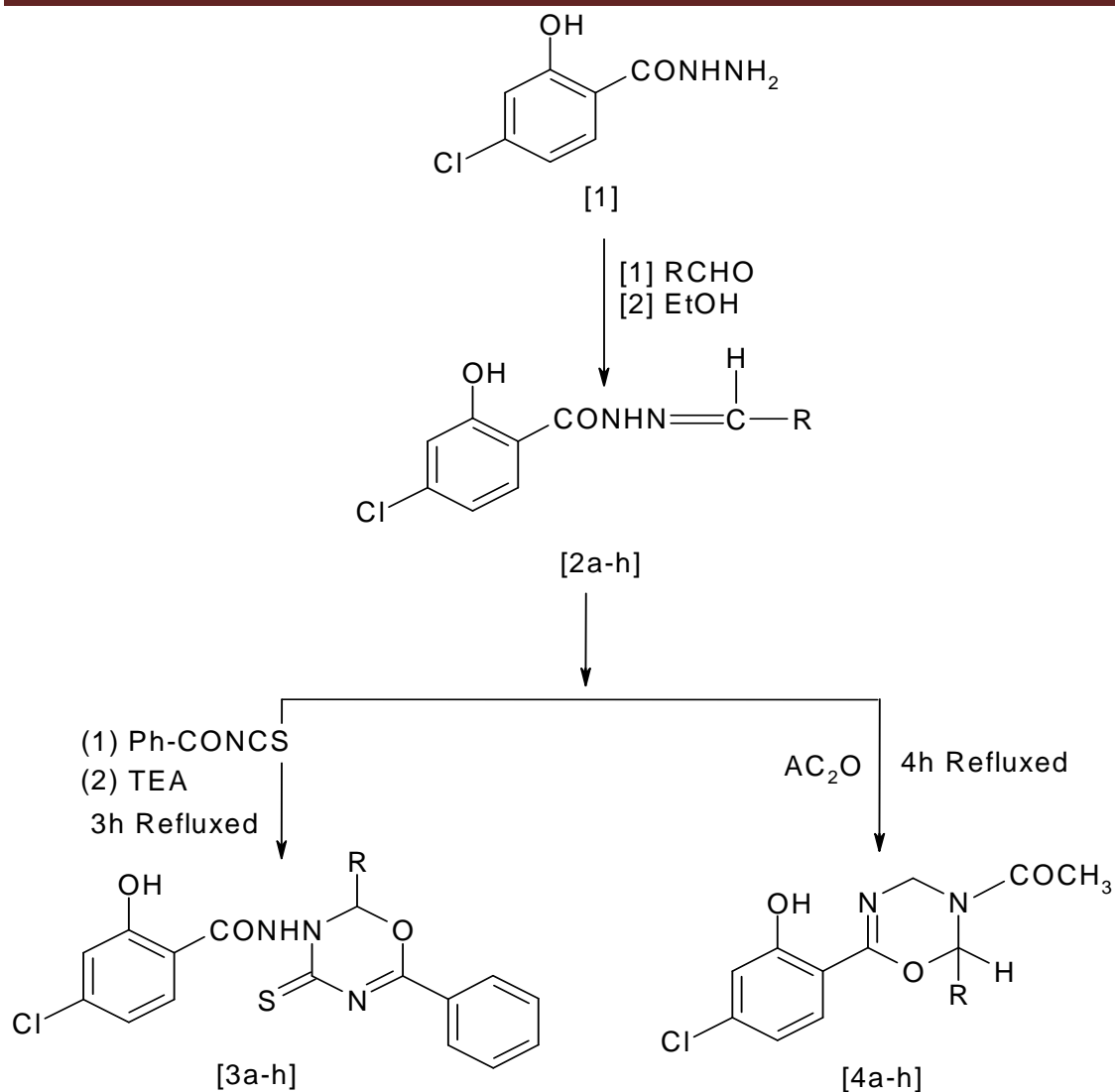
The structures assigned to 1-(4-chloro-2-hydroxybenzamido)-5-oxo-2-arylpyrrolidine-3-carboxylic acid (4a-h) were supported by the elemental analysis and IR spectra showing an absorption bands at 1667-1690cm⁻¹ (C=O of -COOH), 1715-1720 (C=O of pyrrole-2-one), 3030-3080 cm⁻¹ (C-H, of Ar.), 3240-3260 cm⁻¹ (-OH). ¹H NMR: 6.65-7.85 (8H, m) (Ar-H), 5.52-5.60 (1H, s) (-C₅H), 3.33-3.36 (1H, s) (-C₄H), 2.50-2.54, 2.75-2.80 (2H, s) (-C₃H), 6.44-6.68 (1H-s) (-CH-oxadiazole), 11.70-11.85 (1H, s) (-OH), 12.90-12.95 (1H, s)

(-COOH), 4e; 2.38 (3H, s) (-CH₃), 4b, 4g; 3.86 (3H, s) (-OCH₃), 4h; 4.06, (4H, q) (-CH₂), 1.35 (6H, t) (-CH₃), 4f; 6.08 (2H, s) (-OCH₂O cyclic). ¹³C NMR: 110-161 (Aromatic), 50.0-57.0 (-CH of the ring), 37.0-37.5 (-CH₂), 172.0-172.5 (-CO), 178.0-179 (-COOH), (4b,4g): 55.5-56.5 (-OCH₃); (4e): 21.6 (CH₃); (4f): 101.5 (OCH₂O); (4h): 65.3 (OCH₂), 15.0 (CH₃). The C, H, and N analysis data of all compounds are presented in Table-3.

The examination of elemental analytical data reveals that the elemental contents are consistence with the predicted structure shown in Scheme-1. The IR data also direct for assignment of the predicted structure. The final structure of all compounds is confirmed by LC-MS. LC-MS of 3e and 4d compounds are 391 and 340 respectively.

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SCHEME – 1

Where, R =

(a) C₆H₅ (b) 4-OCH₃-C₆H₄ (c) 4-OH-C₆H₄ (d) 2-OH-C₆H₄

(e) 4-CH₃-C₆H₄ (f) 3,4-CH₂O₂-C₆H₄ (g) 4-OH-3-OCH₃-C₆H₃ (h) 3,4-C₂H₅-C₆H₄

Table:-1 Analytical Data and Elemental Analysis of Compounds (2a-h)

Compd.	Molecular formula (Mol.wt.)	Yield	M.P.* °C	Elemental Analysis					
				%C		%H		%N	
				Found	Calcd.	Found	Calcd.	Found	Calcd.
2a	C ₁₄ H ₁₁ ClN ₂ O ₂ (274)	85	243	61.18	61.21	3.99	4.04	10.15	10.20
2b	C ₁₅ H ₁₃ ClN ₂ O ₃ (304)	80	246	59.08	59.12	4.25	4.30	9.14	9.19
2c	C ₁₄ H ₁₁ ClN ₂ O ₃ (290)	75	240	57.79	57.84	3.77	3.81	9.58	9.64
2d	C ₁₄ H ₁₁ ClN ₂ O ₃ (290)	81	243	57.78	57.84	3.75	3.81	9.57	9.64
2e	C ₁₅ H ₁₃ ClN ₂ O ₂ (288)	79	244	62.36	62.40	4.51	4.54	9.64	9.70
2f	C ₁₅ H ₁₁ ClN ₂ O ₄ (318)	75	247	56.49	56.53	3.44	3.48	8.73	8.79
2g	C ₁₅ H ₁₃ ClN ₂ O ₃ (320)	77	249	56.14	56.17	3.04	4.09	8.68	8.73
2h	C ₁₈ H ₁₉ ClN ₂ O ₄ (362)	73	261	59.55	59.59	5.24	5.28	7.68	7.72

* Uncorrected

Table:-2 Analytical Data and Elemental Analysis of Compounds (3a-h)

Compd.	Molecular formula (Mol.wt.)	Yield	M.P.* °C	Elemental Analysis					
				%C		%H		%N	
				Found	Calcd.	Found	Calcd.	Found	Calcd.
3a	C ₂₂ H ₁₇ ClN ₃ O ₃ S (372)	64	245	70.90	70.97	4.50	4.57	11.20	11.29
3b	C ₂₃ H ₁₉ ClN ₃ O ₄ S (402)	65	239	68.61	68.66	4.71	4.73	10.41	10.45
3c	C ₂₂ H ₁₇ ClN ₃ O ₄ S (388)	61	214	67.97	68.04	4.31	4.38	10.78	10.82
3d	C ₂₂ H ₁₇ ClN ₃ O ₄ S (388)	63	216	67.97	68.04	4.31	4.38	10.79	10.82
3e	C ₂₃ H ₁₇ ClN ₃ O ₃ S (386)	60	208	71.47	71.50	4.29	4.40	10.81	10.88

3f	C ₂₃ H ₁₇ ClN ₃ O ₇ S (416)	58	202	66.29	66.35	4.01	4.09	10.01	10.10
3g	C ₂₃ H ₁₇ ClN ₃ O ₅ S (418)	55	204	65.96	66.03	4.01	4.07	10.00	10.05
3h	C ₂₆ H ₂₅ ClN ₃ O ₅ S (460)	59	221	67.79	67.83	5.39	5.43	9.08	9.13

* Uncorrected

Table:-3 Analytical Data and Elemental Analysis of Compounds (4a-h)

Compd.	Molecular formula (Mol.wt.)	Yield	M.P. °C	Elemental Analysis					
				%C		%H		%N	
				Found	Calcd.	Found	Calcd.	Found	Calcd.
4a	C ₁₆ H ₁₄ ClN ₂ O ₃ (317)	63	215	60.51	60.57	4.39	4.42	8.78	8.83
4b	C ₁₇ H ₁₆ ClN ₂ O ₄ (347)	65	218	58.71	58.78	4.58	4.61	8.01	8.07
4c	C ₁₆ H ₁₄ ClN ₂ O ₄ (333)	68	202	57.61	57.66	4.15	4.20	8.37	8.41
4d	C ₁₆ H ₁₄ ClN ₂ O ₄ (333)	66	205	57.59	57.66	4.17	4.20	8.38	8.41
4e	C ₁₇ H ₁₆ ClN ₂ O ₃ (331)	61	219	61.62	61.63	4.79	4.83	8.38	8.46
4f	C ₁₇ H ₁₄ ClN ₂ O ₅ (361)	59	221	56.49	56.51	3.81	3.87	7.69	7.75
4g	C ₁₇ H ₁₆ ClN ₂ O ₅ (363)	57	216	56.15	56.20	4.38	4.41	7.66	7.71
4h	C ₂₀ H ₂₂ ClN ₂ O ₅ (405)	58	224	59.21	59.26	5.36	5.43	6.85	6.91

* Uncorrect

Table:-4 Antibacterial Activity of Compounds (3a-h)

Compounds	Gram +Ve		Gram -Ve	
	Staphylococcus aureus	Bacillus subtilis	E.coli	Klebsiella promioe
3a	12	11	14	13
3b	16	13	15	16
3c	11	12	12	13
3d	13	17	13	18
3e	16	14	15	14
3f	13	20	18	18
3g	12	13	11	15
3h	18	14	14	17
Tetracycline	22	21	19	21

Table:-5 Antifungal Activity of Compounds (3a-h)

Zone of Inhibition at 1000 ppm (%)					
Compounds	Nigrospora Sp.	Aspergillus Niger	Botrydepladia Thiobromine	Rhizopus Nigricum	Fusarium oxyporium
3a	63	63	61	61	62
3b	58	60	62	63	64
3c	61	65	65	64	69
3d	66	63	61	65	70
3e	64	62	59	69	69
3f	59	61	61	68	67
3g	57	65	68	61	65
3h	58	69	70	71	66

Table:-6 Antibacterial Activity of Compounds (4a-h)

Compounds	Gram +Ve		Gram -Ve	
	Staphylococcus aureus	Bacillus subtilis	E.coli	Klebsiella promioe
4a	11	11	12	14
4b	13	12	15	11
4c	15	15	16	12
4d	13	16	15	14
4e	14	14	13	16
4f	16	12	11	14
4g	18	19	18	19
4h	16	15	16	18
Tetracycline	20	21	20	22

Table:-7 Antifungal Activity of Compounds (4a-h)

Zone of Inhibition at 1000 ppm (%)					
Compounds	Nigrospora Sp.	Aspergillus Niger	Botrydepladia Thiobromine	Rhizopus Nigricum	Fusarium oxyporium
4a	61	65	60	59	61
4b	62	64	61	61	63
4c	60	68	63	63	67
4d	61	61	62	61	63
4e	63	63	61	69	64
4f	62	58	64	62	67
4g	67	61	67	59	65
4h	61	63	71	70	67

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